Molecular-based Antiferromagnetic Conductor: β "-(BEDT-TTF)₃[Cu₂(μ -C₂O₄)(C₂O₄)₂(CH₃OH)(H₂O)]

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The organic-inorganic hybrid β "-(BEDT-TTF)₃[Cu₂(μ -C₂O₄)(C₂O₄)₂(CH₃OH)(H₂O)] (BEDT-TTF = bis(ethylenedithio)tetrathia-fulvalene) composed of BEDT-TTF donor and the oxalate-bridged binuclear anion [Cu₂(μ -C₂O₄)(C₂O₄)₂(CH₃OH)(H₂O)]²⁻ has been obtained by electrocrystallization. It crystallizes in the triclinic P 1 space group with cell parameters of a = 7.4803(3) Å, b = 9.3547(3) Å, c = 18.6711(7) Å, α = 95.797(3)°, β = 90.974(3)°, γ = 93.508(3)°, V = 1297.06(8) Å³, and Z = 1 at 100 K. The donor arrangement belongs to the β " phase. From the TTF core bond lengths and Raman spectroscopy, the oxidation state of BEDT-TTF is assigned to ~+2/3. CH₃OH or H₂O molecules bond to the metal atoms at the apical position of the square pyramid with occupancy of 0.5. A supramolecular square lattice forms through hydrogen bonds between the antiferromagnetically binuclear anions in the anion sheet. From band structure at 100 K, metallic conductivity is expected, which agrees with four-probe conductivity measurement: its conductivity is 11.5 S/cm at room temperature, increases to 160 S/cm at 7.6 K, then decreases to 150 S/cm at 2 K. From magnetic measurement, there is no long-range magnetic ordering, which is confirmed by specific heat measurement.

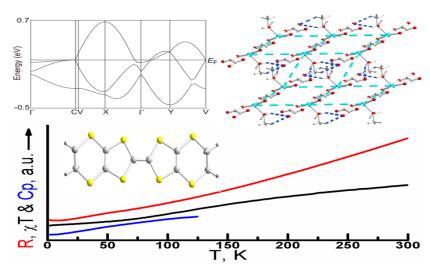


Figure 1. Band structure and anion arrangement in anionic sheet. Temperature-dependent R, χT and Cp.

References

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